

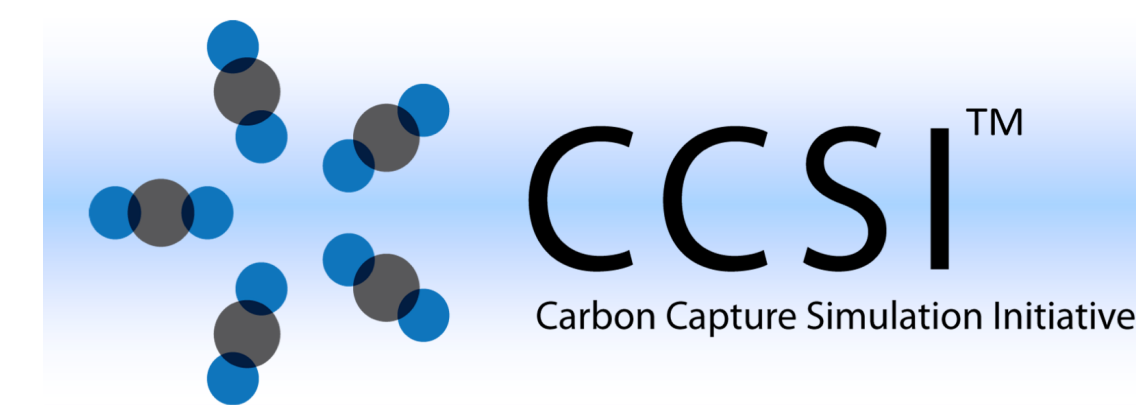
Multi-Scale Modeling with Dynamic Discrepancy

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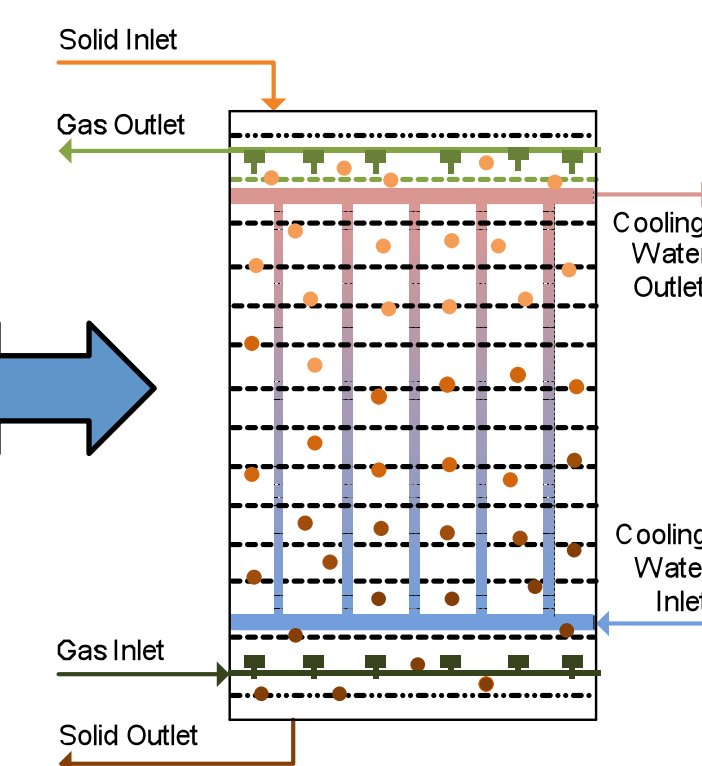
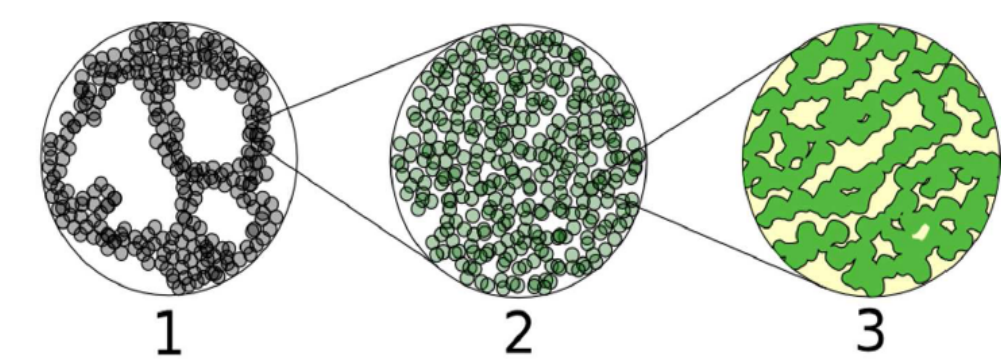
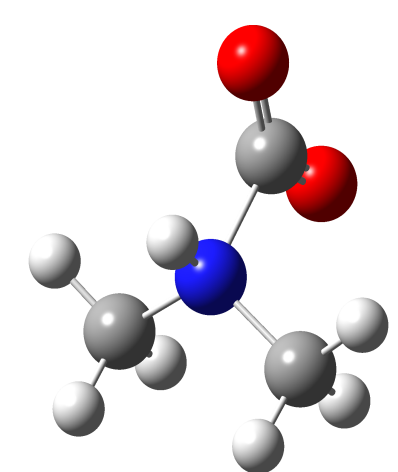
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Multi-Scale Modeling in chemical process systems is an inherently statistical problem. Models for the chemistry at the scale of a catalyst or CO₂ sorbent are incorporated into process-level models. Almost invariably, some information must be left out at the larger scale, leading to uncertainty. Obtaining a good process-scale model means quantifying and minimizing that uncertainty.



What the network map shows is that, if we are only interested in a subset of states in a chemical model, we can usually find some dynamic system containing only those states (with perhaps a few extras) that will provide the correct dynamic behavior.

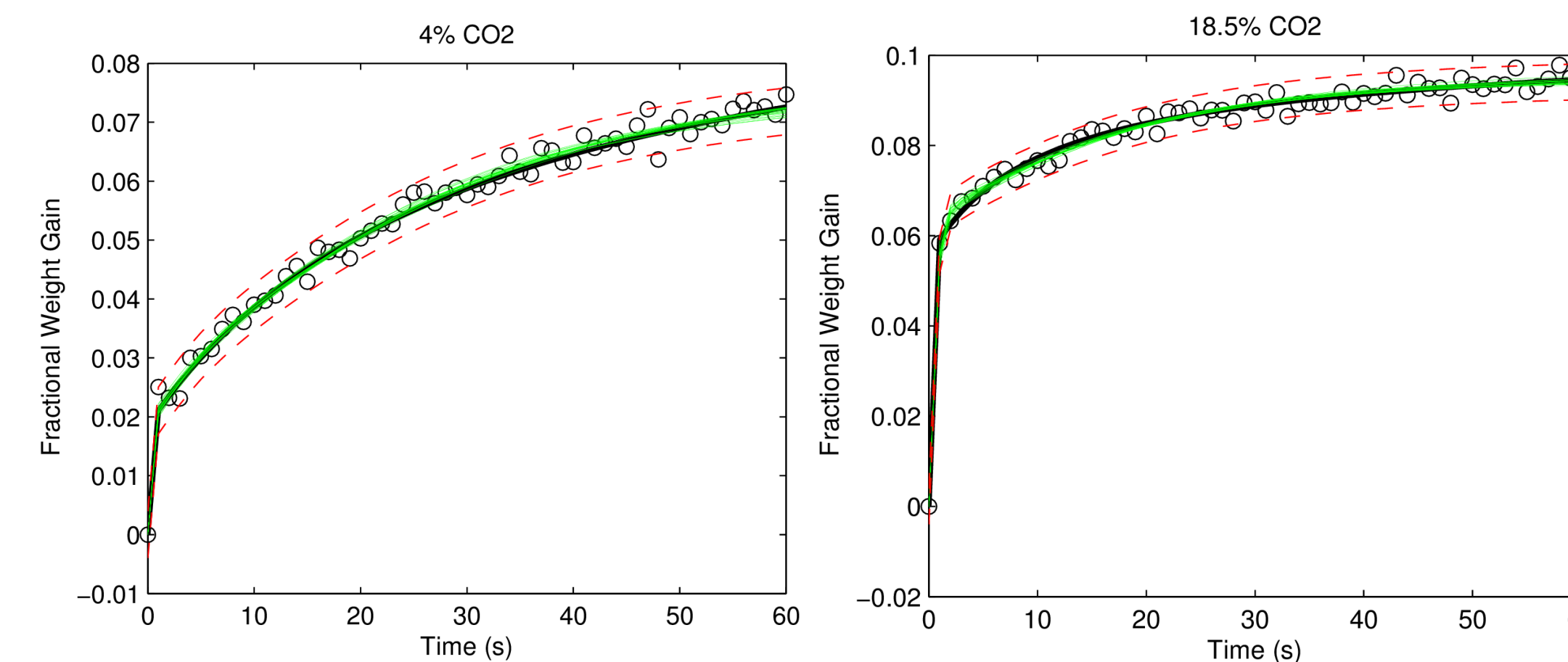
The reduced dynamic system will be found through **Bayesian Nonparametric Regression**. We introduce a stochastic “discrepancy” function δ which is a Gaussian process:²

$$\delta \sim MVN(0, \Gamma)$$

where the covariance kernel Γ may be decomposed as:²

$$\Gamma(\vartheta, \vartheta') = \sigma_0^2 + \sum_{k=1}^K \sigma_k^2 \Gamma_1(\vartheta_k, \vartheta'_k) + \sum_{k=1}^{K-1} \sum_{l=k}^K \sigma_{kl}^2 \Gamma_2([\vartheta_k, \vartheta_l], [\vartheta'_k, \vartheta'_l]) + \dots$$

Simulated TGA data generated for a hypothetical CO₂ sorbent, with draws (green) from the discrepancy model distribution.³



Uncertainty Quantification and Model Reduction are combined in a novel Bayesian approach called **Dynamic Discrepancy**. Consider a dynamic system $\dot{\mathbf{x}} = \mathbf{f}(\mathbf{x}; \mathbf{k}, \boldsymbol{\kappa})$

where \mathbf{x} is a vector of state variables and \mathbf{k} and $\boldsymbol{\kappa}$ are kinetic and equilibrium parameters, respectively. A causal relationship between two states x_i and x_j is established when

$$\dot{x}_i = f_i(\dots, x_j, \dots; k_i, \kappa_i) \Rightarrow x_j \rightarrow x_i$$

Takens has shown that the infinite timeseries of x_i will contain all of the information about the timeseries of x_j .¹

with ϑ a functional input such as a reactant or temperature. The covariance is thus broken down in terms of interactions among inputs: first-order, second-order, etc. The kernels are defined as:²

$$\Gamma_1(\vartheta_k, \vartheta'_k) = \mathcal{B}_1(\vartheta_k)\mathcal{B}_1(\vartheta'_k) + \mathcal{B}_2(\vartheta_k)\mathcal{B}_2(\vartheta'_k) - \frac{1}{4!}\mathcal{B}_4(|\vartheta_k - \vartheta'_k|)$$

$$\Gamma_2([\vartheta_k, \vartheta_l], [\vartheta'_k, \vartheta'_l]) = \Gamma_1(\vartheta_k, \vartheta'_k)\Gamma_1(\vartheta_l, \vartheta'_l)$$

with \mathcal{B}_i the i th Bernoulli polynomial. A Karhunen-Loève decomposition then leads to:²

$$\delta(\vartheta; \boldsymbol{\beta}) = \beta_0 + \sum_{k=1}^K \sum_{m=1}^M \beta_{mk} \varphi_{m1}(\vartheta_k) + \sum_{k=1}^{K-1} \sum_{l=k+1}^K \sum_{m=1}^M \beta_{mkl} \varphi_{m2}(\vartheta_k, \vartheta_l) + \dots$$

here the basis functions φ are deterministic and known, while coefficients $\boldsymbol{\beta}$ are independently distributed:

$$\beta_{mk} \sim N(0, \lambda_{m1} \sigma_k^2) \quad \text{with } \lambda_{ij} \text{ the eigenvalue corresponding to the basis function } \varphi_{ij}$$

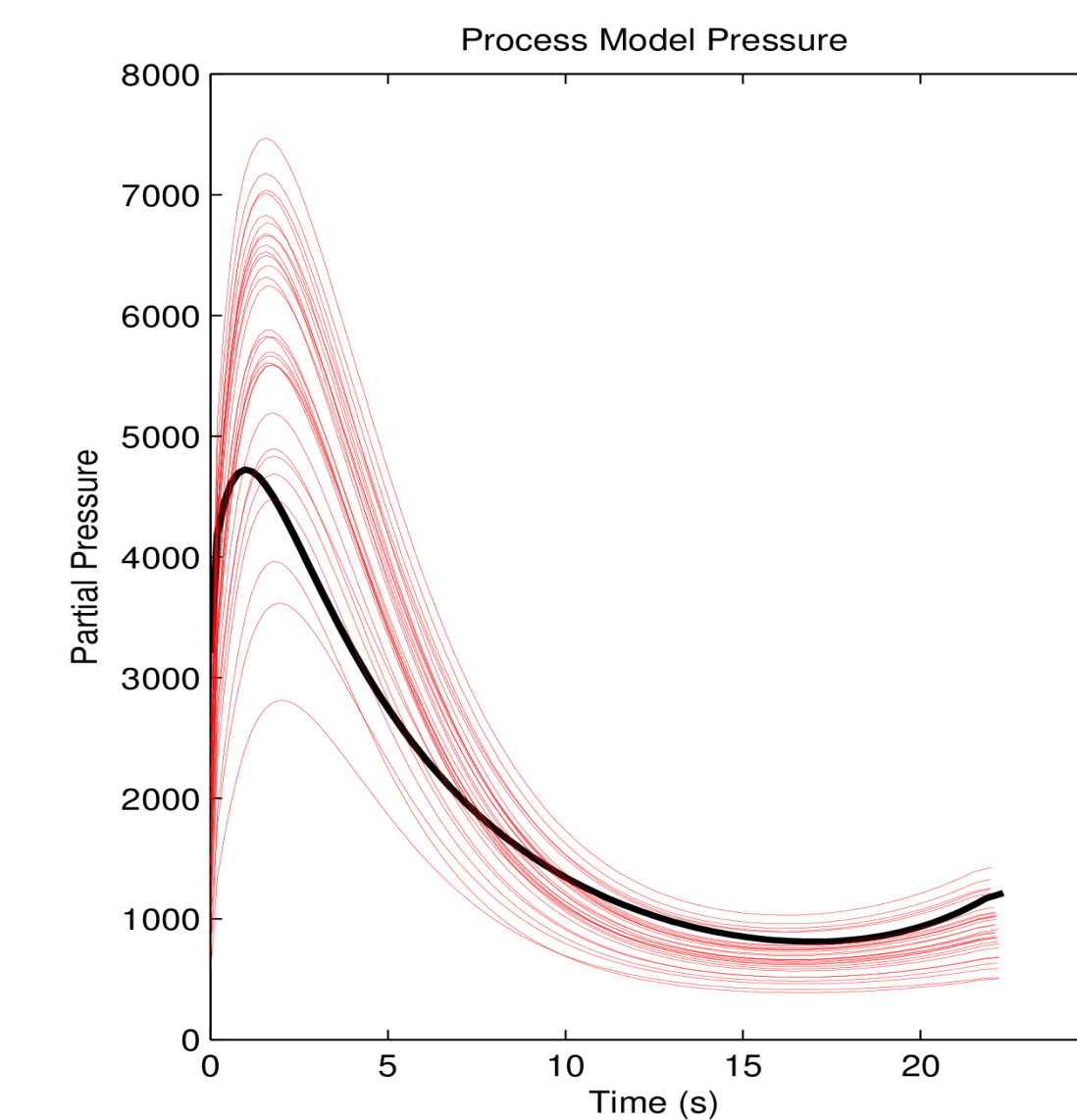
$$\beta_{mkl} \sim N(0, \lambda_{m2} \sigma_{kl}^2) \quad \dot{x}_i = f_i(x_i, x_j; k_i, \kappa_i)$$

$$k_i = k_i^0 \exp[\delta_{k_i}(x_i, x_j, T; \beta_{k_{ij}T})]$$

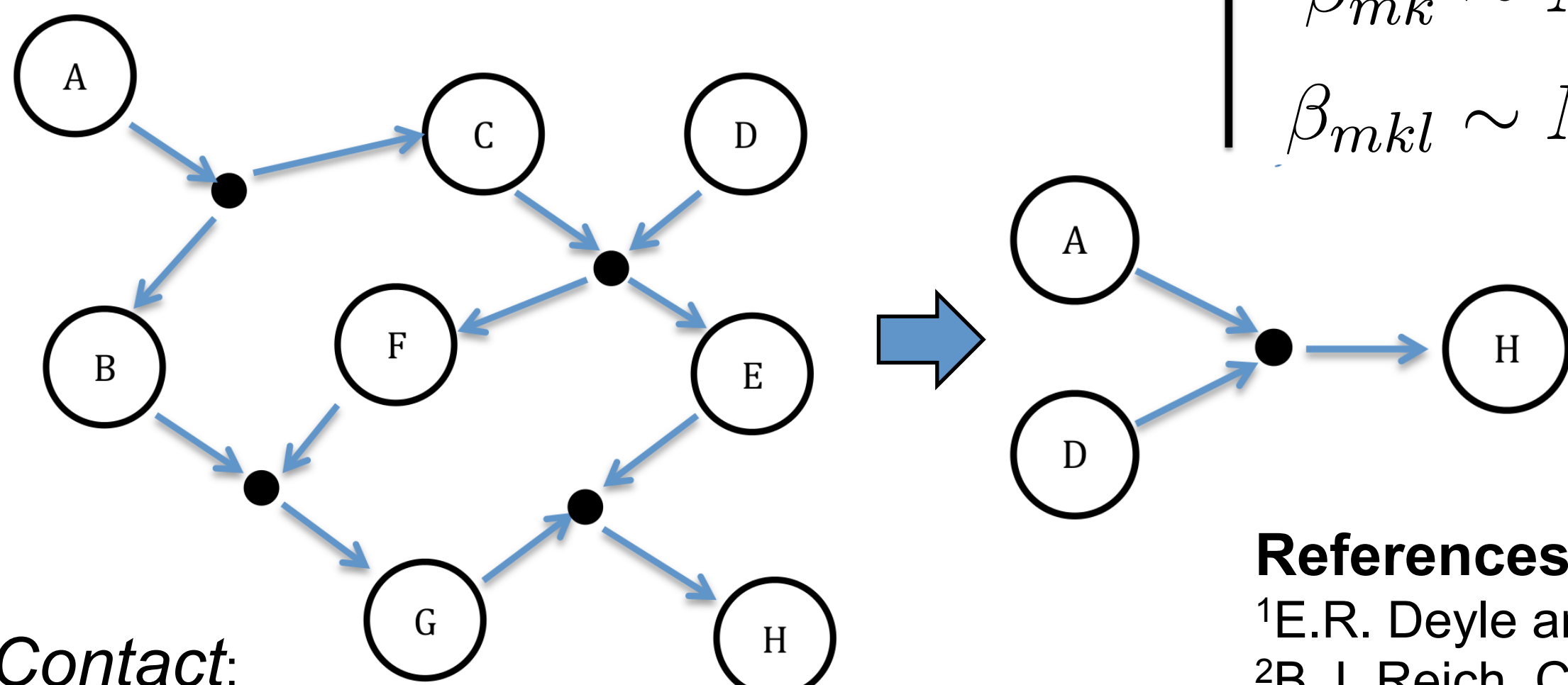
$$\kappa_i = \kappa_i^0 \exp[\delta_{\kappa_i}(x_i, x_j, T; \beta_{\kappa_{ij}T})]$$

The reduced dynamic system is then constructed from the reduced set of states and discrepancy functions as shown at left.³

Uncertainty propagation to the process scale for a hypothetical CO₂ sorbent. Model runs are in red; the “real” process behavior is the black curve. Units are Pa; time is convertible to bed depth.³



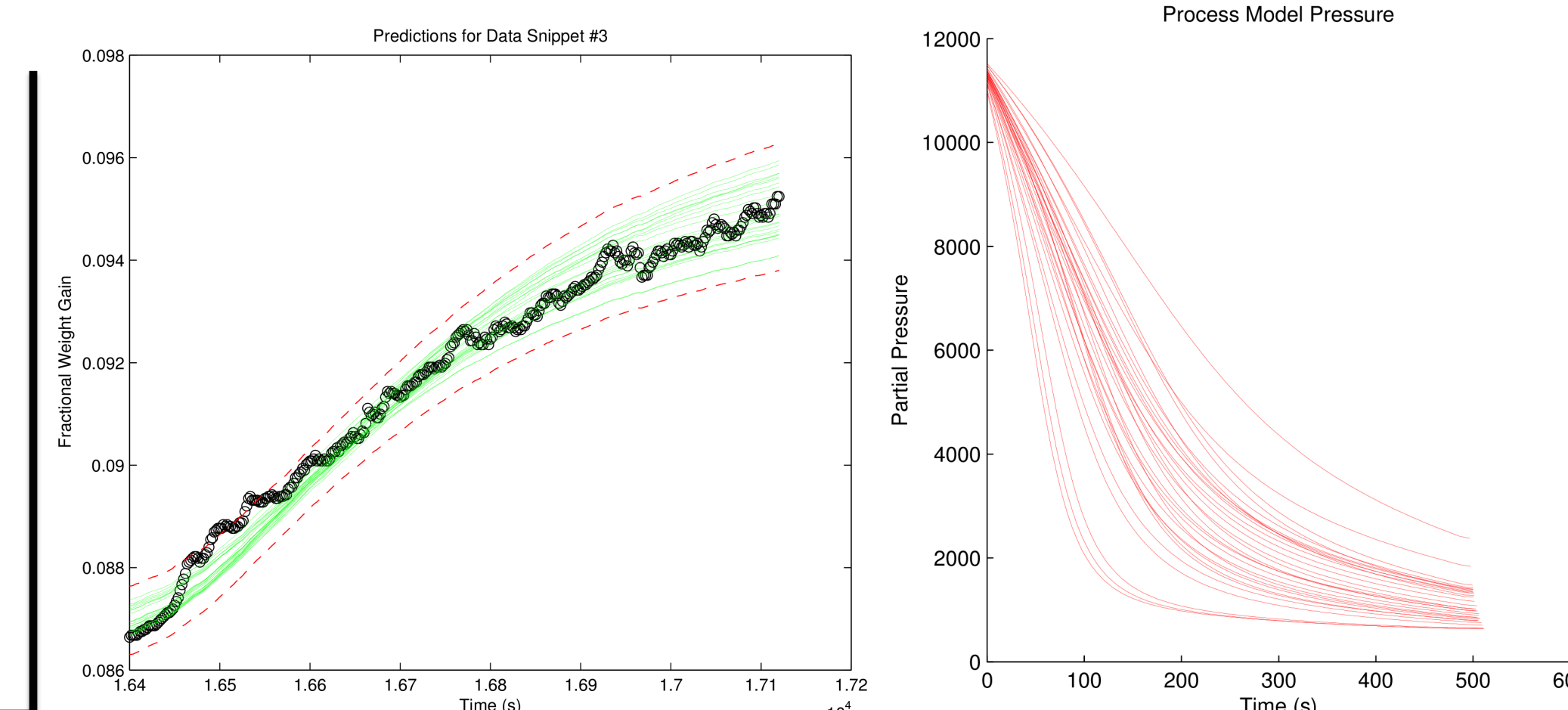
This suggests a method of reducing the order of a reaction network, based on its network topology. In the reaction network below, (circles and letters are states and dots are reactions or other interactions) intermediate states B, C, F, G and E – which propagate causality from the ultimate reactants A and D to the ultimate product H.



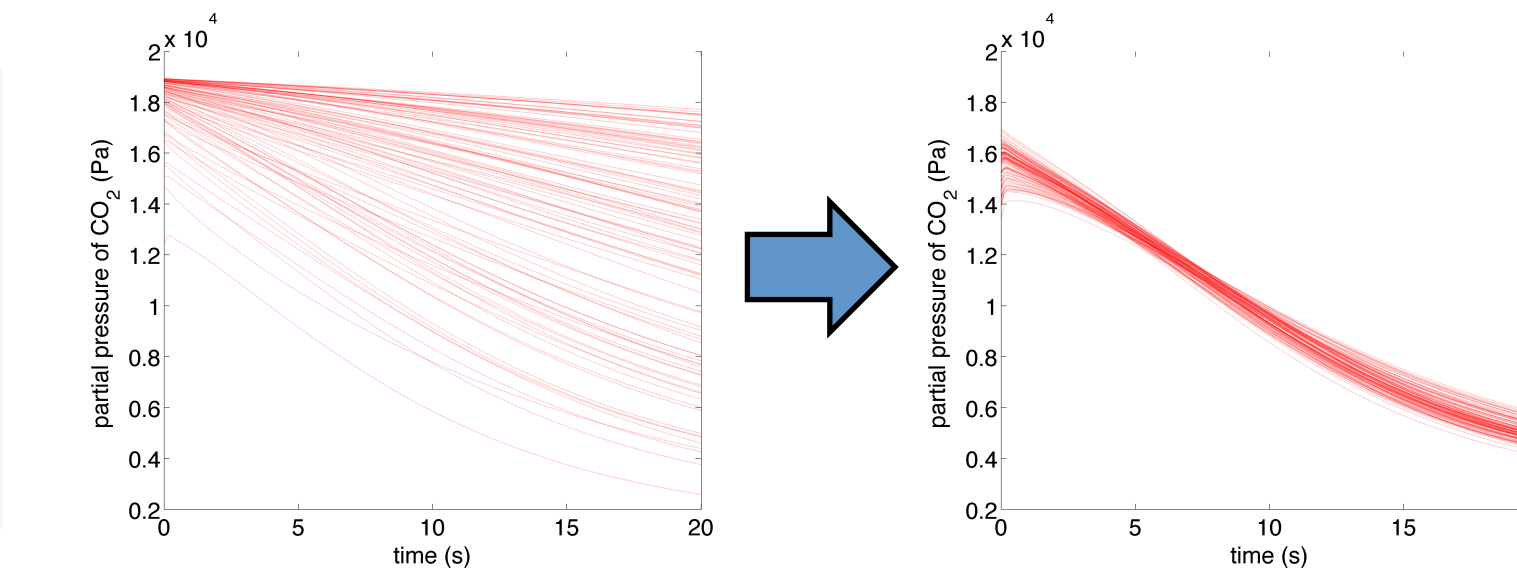
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1. E.R. Deyle and G. Sugihara, PLoS One 6 (2011) e18295.
2. B.J. Reich, C. B. Storlie and H.D. Bondell, Technometrics 51 (2009) 110.
3. K.S. Bhat, D.S. Mebane, C.B. Storlie, P. Mahapatra, submitted.
4. A. Lee and D.C. Miller, Ind. Eng. Chem. Res. 52 (2013) 469.

Above: application of the method to a real PEI-silica sorbent (measurement by James Hoffman, NETL). Left: TGA data calibration. Right: process model upscaling.³



Dynamic Discrepancy is a foundation for machine learning in process design and control. Quantification of uncertainty leads to methods that can minimize uncertainty.



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